# Stereodivergent Total Syntheses of Precoccinelline, Hippodamine, Coccinelline, and Convergine ${ }^{\dagger}$ 

Aleksey I. Gerasyuto and Richard P. Hsung*<br>Division of Pharmaceutical Sciences and Department of Chemistry, Rennebohm Hall, 777 Highland Avenue, University of Wisconsin, Madison, Wisconsin 53705<br>rhsung @ wisc.edu

Received August 4, 2006

## ABSTRACT



A stereodivergent approach toward total syntheses of Coccinellidae defensive alkaloids is described. These syntheses feature a highly diastereoselective intramolecular aza-[3+3] annulation strategy, which represents a de novo approach to this family of natural products.

Ladybird beetles (Coccinellidae) play an important role in controlling populations of agricultural pests such as aphids, mealy bugs, and scale insects, and to protect themselves from their natural predators such as ants and quails, they utilize a reflex bleeding mechanism. ${ }^{1}$ When scarred or disturbed, they release an orange fluid from their joints that contains a mixture of defensive alkaloids. ${ }^{2}$ In 1971, Tursch and coworkers ${ }^{3 a}$ first isolated precoccinelline $\mathbf{1}$ from this fluid along with a crystalline substance, which was confirmed as the N -oxide of $\mathbf{1}$ or as coccinelline $\mathbf{2}$ by X-ray analysis. ${ }^{3 b}$ Isolations of hippodamine ${ }^{3 c, d} \mathbf{3}$, its $N$-oxide convergine ${ }^{3 \mathrm{c}, \mathrm{d}}$ 4, and the thermodynamically most stable all syn isomer myrrhine $^{3 \mathrm{e}} \mathbf{5}$ were reported later. Alkaloids $\mathbf{1}$ and $\mathbf{3}$ represent

[^0]the two other possible stereoisomers of the 2-methyl-perhydro- 9 b -azaphenalene system with the methyl group being equatorial. Interestingly, $N$-oxide of myrrhine 5 does not occur naturally.


1: precoccinelline
2: N -oxide: coccinelline


3: hippodamine
4: N -oxide: convergine


| $\overline{\mathrm{H}}$ |
| :---: |
| I |



5: myrrhine

Their unique structural feature has attracted a lot of attention ${ }^{4-6}$ in the last 30 years, commencing with Ayer's first total syntheses of all these alkaloids in $1976^{4 a, 5 a}$ and some elegant constructions of $\mathbf{1}$ and $\mathbf{2}$ by Mueller, ${ }^{5 b, c}$ who also synthesized $\mathbf{3}-\mathbf{5},{ }^{5 \mathrm{c}}$ and by Stevens in 1979. ${ }^{5 \mathrm{~d}}$ Interestingly, most of these syntheses intercept a key intermediate

Scheme 1. Retrosynthetic Analysis

in Ayer's syntheses, ${ }^{5 \mathrm{a}}$ and in all syntheses of these alkaloids except for one, ${ }^{4 a}$ the equatorial methyl group is being introduced in the latter steps of the sequence. Herein, we report a de novo approach to precoccinelline $\mathbf{1}$ and hippodamine $\mathbf{3}$ and their respective $N$-oxides 2 and 4, featuring a highly stereoselective intramolecular aza-[3+3] annulation. ${ }^{7-10}$
(4) For syntheses of hippodamine, convergine, and myrrhine, see: (a) Ayer, W. A.; Dawe, R.; Eisner, R. A.; Furuichi, K. Can. J. Chem. 1976, 54, 473. (b) Adams, D. R.; Carruthers, W.; Crowley, P. J. J. Chem. Soc., Chem. Commun. 1991, 1261. (c) Rejzek, M.; Stockman, R. A.; Hughes, D. L. Org. Biomol. Chem. 2005, 3, 73.
(5) For syntheses of precoccinelline and coccinelline, see: (a) Ayer, W. A.; Furuichi, K. Can. J. Chem. 1976, 54, 1494. (b) Mueller, R. H.; Thompson, M. E. Tetrahedron Lett. 1979, 1991. (c) Mueller, R. H.; Thompson, M. E.; DiPardo, R. M. J. Org. Chem. 1984, 49, 2271. (d) Stevens, R. V.; Lee, A. W. M. J. Am. Chem. Soc. 1979, 101, 7032. (e) Yue, C.; Nicolay, J.-F.; Royer, J.; Husson, H.-P. Tetrahedron 1994, 50, 3139. (f) Takahata, H.; Ouchi, H.; Ichinose, M.; Nemoto, H. Org. Lett. 2002, 4, 3459.
(6) For an elegant total synthesis of related alkaloid (-)-205B, see: Smith, A. B., III.; Kim, D. S. Org. Lett. 2005, 7, 3247.
(7) For reviews, see: (a) Harrity, J. P. A.; Provoost, O. Org. Biomol. Chem. 2005, 3, 1349. (b) Hsung, R. P.; Kurdyumov, A. V.; Sydorenko, N. Eur. J. Org. Chem. 2005, 23. (c) Coverdale, H. A.; Hsung, R. P. ChemTracts 2003, 16, 238.
(8) For a review on vinylogous amide chemistry, see: Kucklander, U. Enaminones as Synthons. In The Chemistry of Functional Groups: The Chemistry of Enamines Part I. Rappoport, Z., Ed. John Wiley \& Sons: New York, 1994; p 523.
(9) For recent studies in this area, see: (a) Pattenden, L. C.; Wybrow, R. A. J.; Smith, S. A.; Harrity, J. P. A. Org. Lett. 2006, 8, 3089. (b) Shintani, R.; Hayashi, T. J. Am. Chem. Soc. 2006, 128, 6330. (c) Halliday, J. I.; Chebib, M.; Turner, P.; McLeod, M. D. Org. Lett. 2006, 8, 3399. (d) Bose, D. S.; Kumar, R. K. Heterocycles 2006, 68, 549. (e) Goodenough, K. M.; Raubo, P.; Harrity, J. P. A. Org. Lett. 2005, 7, 2993. (f) Goodenough, K. M.; Moran, W. J.; Raubo, P.; Harrity, J. P. A. J. Org. Chem. 2005, 70, 207. (g) Agami, C.; Dechoux, L.; Hebbe, S.; Ménard, C. Tetrahedron 2004, 60, 5433. (h) Ji, S.-J.; Jiang, Z.-Q.; Lu, J.; Loh, T.-P. Synlett 2004, 831. (i) Hedley, S. J.; Moran, W. J.; Price, D. A.; Harrity, J. P. A. J. Org. Chem. 2003, 68, 4286.
(10) For intramolecular aza-[3+3] annulation, see: (a) Swidorski, J. J.; Wang, J.; Hsung, R. P. Org. Lett. 2006, 8, 777. (b) Gerasyuto, A. I.; Hsung, R. P.; Sydorenko, N.; Slafer, B. W. J. Org. Chem. 2005, 70, 4248. (c) Sydorenko, N.; Zificsak, C. A.; Gerasyuto, A. I.; Hsung, R. P. Org. Biomol. Chem. 2005, 3, 2140. (d) Luo, S.; Zificsak, C. Z.; Hsung, R. P. Org. Lett. 2003, 5, 4709. (e) Wei, L.-L.; Sklenicka, H. M.; Gerasyuto, A. I.; Hsung, R. P. Angew. Chem., Int. Ed. 2001, 40, 1516. For references on intermolecular aza-[3 + 3] annulation, see: (f) Sydorenko, N.; Hsung, R. P.; Vera, E. L. Org. Lett. 2006, 8, 2611. (g) Sydorenko, N.; Hsung, R. P.; Darwish, O. S.; Hahn, J. M.; Liu, J. J. Org. Chem. 2004, 69, 6732. (h) Sklenicka, H. M.; Hsung, R. P.; McLaughlin, M. J.; Wei, L.-L.; Gerasyuto, A. I.; Brennessel, W. W. J. Am. Chem. Soc. 2002, 124, 10435.

Retrosynthetically, as shown in Scheme 1, we envisioned that both $\mathbf{1}$ and $\mathbf{3}$ could be derived from the same aza-tricyclic intermediate $\mathbf{A}$ upon hydrogenation of the double bonds and removal of the carboxymethyl group. The key tricycle A can be prepared via an intramolecular aza-[3+3] annulation of vinylogous urethane $\mathbf{6},{ }^{7,10}$ which should be accessible in several steps from cis-1,3-disubstituted lactam 7 with bromide 9 and glutarimide $\mathbf{8}$ as the essential starting points.

Our synthesis commenced with alkylation of TBDPSprotected propargyl alcohol $\mathbf{1 0}^{11,12}$ employing excess $1,3-$ dibromopropane followed by Lindlar hydrogenation that led to bromide 9 in 59\% overall yield (Scheme 2). Lactam 7

was prepared via reductive alkylation ${ }^{13}$ of 4-methyl glutarimide 8. ${ }^{14}$ The Grignard reagent generated from bromide 9 was added to the Mg salt $\mathbf{8 a}$ formed in situ from glutarimide $\mathbf{8}$ and 1.0 equiv of $\mathrm{CH}_{3} \mathrm{MgCl}$, and after stirring overnight at room temperature, $\mathrm{NaBH}_{3} \mathrm{CN}$ and AcOH were added to reduce the intermediate hemi-aminal (see 11a). The reduction proceeded stereoselectively and afforded lactam 7 in $75 \%$ overall yield exclusively as the 1,3-syn isomer. This is likely a result of an axial approach of hydride to the conformation shown for $N$-acyliminium ion $\mathbf{1 1 b}$.

Converting 7 to the corresponding thiolactam using Lawesson's reagent ${ }^{15}$ followed by alkylation with $\alpha$-bromo methyl acetate gave thiol ether $\mathbf{1 2}$ in $90 \%$ yield over two steps (Scheme 3). Eschenmoser sulfide contraction ${ }^{16}$ of $\mathbf{1 2}$ proceeded smoothly and led to the protected Z-vinylogous

[^1]Scheme 3. Synthesis of Allyl Alcohol 14

urethane 13. Subsequent deprotection of the TBDPS group with TBAF furnished allyl alcohol 14. It is noteworthy that the described sequence was suitable for multigram scale synthesis of alcohol $\mathbf{1 4}$ with an $82 \%$ overall yield starting from lactam 7.
$\mathrm{Pyr} \cdot \mathrm{SO}_{3}$ oxidation ${ }^{17}$ of allyl alcohol 14 led to the corresponding enal $\mathbf{1 5}$ as a 7:1 cis/trans isomeric mixture (Scheme 4). Enal 15 was subjected to the aza- $[3+3]$ annulation

Scheme 4. Key Intramolecular Aza- $[3+3]$ Annulation

conditions employing piperidinium trifluoro acetate in EtOAc. The desired cycloadduct $\mathbf{1 6}$ was obtained as a single diastereomer in $51 \%$ yield, and its anti relative stereochemistry at the ring junction was established using nOe's (see the box in Scheme 4) and by X-ray of a later intermediate. The observed stereoselectivity is in good agreement with our previous experience. ${ }^{10 \mathrm{~b}}$

Because of the precarious nature of 16, we found it more convenient to hydrogenate the reaction mixture over $\mathrm{Pd}(\mathrm{OH})_{2}$

[^2]a.

b.


Figure 1. X-ray structure of the picrate salt of 18b. (a) Structure with picrate. (b) Structure with picrate removed for clarity.
after the annulation. ${ }^{10 e}$ The one-pot protocol allowed us to access the more stable aza-tricycle $\mathbf{1 7}$ with a consistent $43 \%$ overall yield from 15. It is noteworthy that aza-tricycle 17 contains three of the four stereogenic centers required for precoccinelline 1 and hippodamine 3.

With aza-tricycle 17 in hand, we hydrogenated the internal double bond employing Adam's catalyst ${ }^{18}$ (Scheme 5).

Scheme 5. Stereodivergent Hydrogenation of $\mathbf{1 7}$


Hydrogenation occurred in a stereodivergent manner, leading to both isomers 18a and 18b with a ratio of 2:1. Major isomer 18a resembles the aza-tricyclic manifold in precoccinelline $\mathbf{1}$, whereas $\mathbf{1 8 b}$ has the framework of hippodamine 3.

[^3]Although esters 18a,b could be separated by a tedious alumina gel column, it was found that they could be readily resolved via a selective alkaline hydrolysis of the equatorial ester 18a (Scheme 5). Upon treatment of the crude hydrogenation mixture with 1.7 M aq KOH , unreacted axial ester 18b was recovered by simple extraction in $36 \%$ overall yield from 17. Upon acidification of the aqueous phase, acid 19a was isolated in $49 \%$ yield with its assigned relative stereochemistry being supported by nOe's. ${ }^{12}$ The structure of $\mathbf{1 8 b}$ was unambiguously assigned by X-ray analysis of its corresponding picrate salt (Figure 1).

To complete the total synthesis of precoccinelline 1, acid 19a was subjected to Barton's decarboxylation conditions ${ }^{19}$ (Scheme 6). The desired alkaloid was isolated in $43 \%$ overall

yield with all physical data matching the reported literature data. ${ }^{5 c, 21}$ Subsequent oxidation of precoccinelline employing $m$-CPBA provided coccinelline 2 in excellent yield.

The synthesis of hippodamine 3 turned out to be more challenging, as all attempts to directly hydrolyze ester 18a failed. This can be attributed to the inaccessibility of the hindered axial ester group in $\mathbf{1 8 b}$. We attempted to epimerize the axial ester to the more stable equatorial one using $\mathrm{K}_{2} \mathrm{CO}_{3} / \mathrm{MeOH}$ and $\mathrm{DBU} /$ toluene protocols, but neither condition was suitable with complete recovery of the starting material.

Ultimately, we found that treatment of ester $\mathbf{1 8 b}$ with KHMDS ${ }^{20}$ at $-78{ }^{\circ} \mathrm{C}$ with subsequent quenching of the

[^4]reaction mixture with MeOH at $0^{\circ} \mathrm{C}$ led to the desired equatorial ester $\mathbf{1 8 c}$ with $86 \%$ yield (Scheme 7). Hydrolysis

Scheme 7. Completion of Total Syntheses of $\mathbf{3}$ and 4

of $\mathbf{1 8 c}$ was successful using 1.7 M aq KOH at $50^{\circ} \mathrm{C}$, leading to carboxylic acid 19b in $44 \%$ yield. Subsequently, hippodamine 3 was obtained in $43 \%$ overall yield after decarboxylation of 19b under Barton's conditions. ${ }^{19}$ Successive oxidation of $\mathbf{3}$ with $m$-CPBA afforded convergine $\mathbf{4}$ in $88 \%$ yield. Both natural products matched the literature spectroscopic data. ${ }^{4 \mathrm{c}, 21}$

We have described here total syntheses of precoccinelline and hippodamine (overall yields from glutarimide are $4.8 \%$ and $1.3 \%$, respectively) and their respective $N$-oxides coccinelline and convergine (overall yields from glutarimide are $4.6 \%$ and $1.2 \%$, respectively), featuring a stereoselective intramolecular aza- $[3+3]$ annulation strategy, an interesting Eschenmoser sulfide contraction, and a stereodivergent hydrogenation. This work provides a novel approach toward the 2-methyl-perhydro-9b-azaphenalene family of alkaloids.

Acknowledgment. The authors thank the NIH [NS38049] for funding and Mr. Benjiman E. Kucera and Dr. Vic Young from the University of Minnesota for providing X-ray structural analysis. A.I.G. thanks Dr. Lev Lis for valuable discussions. This work was carried out in part at UMN.

Supporting Information Available: Experimental and ${ }^{1} \mathrm{H}$ NMR spectral data and characterizations for all new compounds as well as X-ray structural data. This material is available free of charge via the Internet at http://pubs.acs.org.

OL0619359


[^0]:    ${ }^{\dagger}$ Dedicated to Professor Jeffrey D. Winkler on the occasion of his 50th birthday.
    (1) Happ, G. M.; Eisner, T. Science 1961, 134, 329.
    (2) For reviews, see: (a) King, A. G.; Meinwald, J. Chem. Rev. 1996, 96, 1105. (b) Stevens, R. V. Acc. Chem. Res. 1984, 17, 289.
    (3) (a) Tursch, B.; Daloze, D.; Dupont, M.; Pasteels, J. M.; Tricot, M.C. Experientia 1971, 27, 1380. (b) Karlsson, R.; Losman, D. J. Chem. Soc., Chem. Commun. 1972, 626. (c) Tursch, B.; Daloze, D.; Pasteels, J. M.; Cravador, A.; Braekman, J. C.; Hootele, C.; Zimmermann, D. Bull. Soc. Chim. Belg. 1972, 81, 649. (d) Tursch, B.; Daloze, D.; Braekman, J. C.; Hootele, C.; Cravador, A.; Losman, D.; Karlsson, R. Tetrahedron Lett. 1974, 409. (e) Tursch, B.; Daloze, D.; Braekman, J. C.; Hootele, C.; Pasteels, J. M. Tetrahedron 1975, 31, 1541.

[^1]:    (11) Toshima, K.; Ohta, K.; Ohashi, A.; Nakamura, T.; Nakata, M.; Tatsuta, K.; Matsumura, S. J. Am. Chem. Soc. 1995, 117, 4822.
    (12) See Supporting Information.
    (13) (a) Karstens, W. F. J.; Stol, M.; Rutjes, F. P. J. T.; Kooijman, H.; Spek, A. L.; Hiemstra, H. J. Organomet. Chem. 2002, 624, 244. (b) Esch, P. M.; Hiemstra, H.; Klaver, W. J.; Speckamp, W. N. Heterocycles 1987, 26, 75. (c) Evans, D. A.; Thomas, E. W.; Cherpeck, R. E. J. Am. Chem. Soc. 1982, 104, 3695.
    (14) Handley, G. J.; Nelson, E. R.; Somers, T. C. Aust. J. Chem. 1960, 13, 129.
    (15) For a recent review, see: Jesberger, M.; Davis, T. P.; Barner, L. Synthesis 2003, 1929.
    (16) (a) Roth, M.; Dubs, P.; Götchi, E.; Eschenmoser, A. Helv. Chim. Acta 1995, 54, 710. (b) Shiosaki, K. In Comprehensive Organic Synthesis; Trost, B. M., Ed.; Pergamon: Oxford, 1991; Vol. 2, p 865.

[^2]:    (17) (a) Chen, L.; Lee, S.; Renner, M.; Tian, Q.; Nayyar, N. Org. Proc. Res. Dev. 2006, 10, 163. (b) Parikh, J. R.; Doering, W. von E. J. Am. Chem. Soc. 1967, 89, 5505.

[^3]:    (18) (a) Akhrem, A. A.; Lakhvich, F. A.; Lis, L. G.; Pshenichnyi, V. N.; Arsen'ev, A. S. Zh. Org. Khim. 1980, 16, 1290. (b) Akhrem, A. A.; Lakhvich, F. A.; Lis, L. G.; Kuz'mitskii, B. B.; Mizulo, N. A.; Gorbacheva, I. A. Zh. Org. Khim. 1985, 21, 1348.

[^4]:    (19) (a) Barton, D. H. R.; Bridon, D.; Fernandez-Picot, I.; Zard, S. Z. Tetrahedron 1987, 43, 2733. (b) Barton, D. H. R.; Hervé, Y.; Potier, P.; Thierry, J. Tetrahedron 1987, 44, 5479.
    (20) Klotz, P.; Mann, A. Tetrahedron Lett. 2003, 44, 1927.
    (21) Lebrun, B.; Braekman, J. C.; Daloze, D. Magn. Reson. Chem. 1999, 37, 60.

